

STRUCTURE OF A NEW  
ANTIBACTERIAL ANTIBIOTIC,  
THIOTETROMYCIN

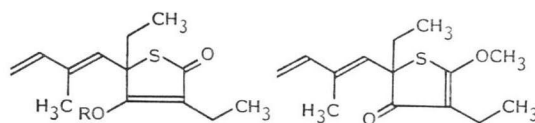
Sir:

In the course of screening for new substances, a new antibacterial antibiotic, thiotetromycin (**1**)<sup>1)</sup>, mp 92°C,  $[\alpha]_D^{25} +124^\circ$  (*c* 1.0, MeOH), C<sub>13</sub>H<sub>13</sub>O<sub>2</sub>S (M<sup>+</sup>, *m/z* 238.102), was found in the culture broth of *Streptomyces* sp. strain OM-674.

We now report structural analysis of **1** by means of NMR spectroscopy. The following data suggested that **1** possesses two chromophore moieties, a conjugated diene and an  $\alpha,\beta$ -unsaturated thiolactone: UV  $\lambda_{\text{max}}^{\text{EtOH}}$  238 ( $\epsilon$  30,100) and 300 nm (4,700) and IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  1620 cm<sup>-1</sup>

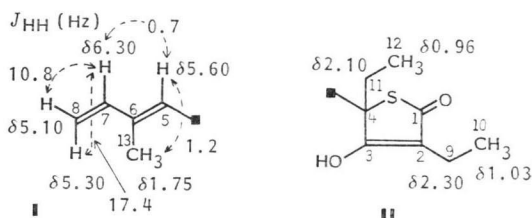
O  
||  
R-S-C-C=C

The <sup>13</sup>C NMR spectrum of **1** suggested the presence of a carbonyl carbon ( $\delta$  198.3), an oxygenated olefinic carbon ( $\delta$  179.2), five olefinic carbons ( $\delta$  141.3, 140.4, 129.4, 118.1 and 113.8), a quaternary carbon ( $\delta$  60.8), two methylenes ( $\delta$  33.3 and 16.1) and three methyls ( $\delta$  12.5, 12.3 and 8.5). The presence of an enol in **1** was confirmed from the IR absorption at 1780 cm<sup>-1</sup> in the monoacetate **2**, which was obtained by acetylation of **1** with Ac<sub>2</sub>O/pyridine. The disappearance of a broad signal at  $\delta$  7.5 in the <sup>1</sup>H NMR spectrum of **1** on addition of D<sub>2</sub>O also confirmed the presence of an enol. The <sup>1</sup>H NMR spectrum indicated the presence of one methyl ( $\delta$  1.75) and one ethyl group ( $\delta$  1.03, 3H, t, *J* = 7.5 Hz and  $\delta$  2.30, 2H, q, *J* = 7.5 Hz) attached to double bond, one ethyl group ( $\delta$  0.96, 3H, t, *J* = 7.2 Hz and  $\delta$  2.10, 2H, q, *J* = 7.2 Hz) attached to a quaternary carbon, four olefinic protons ( $\delta$  5.10, 5.30, 5.60 and 6.30) and one enolic proton. The detailed proton spin decoupling experiment of **1** revealed the presence of a butadienyl moiety [I] containing a methyl group. The validity of the partial structure [I] was also confirmed from the <sup>13</sup>C{<sup>1</sup>H} long range selective proton decoupling (LSPD) experiment of **1**. The methyl group of C-13 in [I] should be located at C-6 from the observation that irradiation of the methyl proton ( $\delta$  1.75) collapsed each of the two olefinic carbons at C-5 ( $\delta$  129.4, broad singlet) and at C-7 ( $\delta$  141.3, multiplet), into a clear doublet ( $^8J_{\text{CH}} = 4.3$  Hz with H-7 for C-5 and  $^8J_{\text{CH}} = 9.3$  Hz with H-5 for C-7, respectively), since the values of  $^2J$  (C-7, H-8) and  $^2J$  (C-5, H-11) are negligibly small. Furthermore, the configuration of the



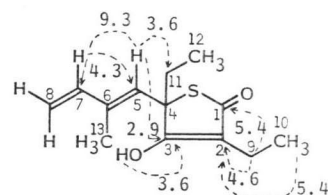
- 1** R = H  
**2** R = COCH<sub>3</sub>  
**3** R = CH<sub>3</sub>

**4**



**I**

**II**



LSPD Pattern [ $^2J_{\text{CH}}$ ,  $^3J_{\text{CH}}$  (Hz)]

Table 1. <sup>13</sup>C NMR data in CDCl<sub>3</sub>.

Carbon No.	Chemical shift (ppm)/TMS		
	<b>1</b>	<b>3</b>	<b>4</b>
1	198.3 (s)*	194.0	185.8
2	118.1 (s)	120.0	114.7
3	179.2 (s)	178.2	201.5
4	60.8 (s)	60.2	67.7
5	129.4 (d)	131.0	131.5
6	140.4 (s)	139.1	139.7
7	141.3 (d)	141.5	141.4
8	113.8 (t)	113.5	113.5
9	33.3 (t)	33.6	34.0
10	12.5 (q)	12.4	13.6
11	16.1 (t)	17.4	16.6
12	8.5 (q)	8.6	8.8
13	12.3 (q)	14.4	12.8
3-OCH <sub>3</sub>		59.6	
1-OCH <sub>3</sub>			59.0

\* Multiplicity: s; singlet, d; doublet, t; triplet, q; quartet.

conjugated diene moiety was found to be identical with that of thiolactomycin<sup>2,3)</sup>, with ethyl groups being replaced by methyl groups in the latter case, from comparison of both NMR spectral data.

The existence of two ethyl groups, a quaternary carbon and a carbonyl and enolic carbons in the remaining portion  $C_8H_{11}O_2S$  and the following LSPD experiment of **1** led us to the 5-membered  $\alpha,\beta$ -unsaturated thiolactone ring [**II**] as another chromophore. Upon irradiation of the methylenic proton ( $\delta$  2.30) at C-9, the carbonyl carbon ( $\delta$  198.3,  $^3J_{CH}=5.4$  Hz) and the olefinic carbon ( $\delta$  118.1,  $^2J_{CH}=4.6$  Hz and  $^3J_{CH}=5.4$  Hz) collapsed to a singlet and a quartet, respectively. This suggested that one of two ethyl groups must be located at the  $\alpha$ -position to the thioester carbonyl. On the other hand, upon irradiation of the olefinic proton at  $\delta$  5.60, the methylenic carbon ( $\delta$  16.1, broad doublet,  $^3J_{CH}=3.6$  Hz) at C-11 and the broad carbon signal ( $\delta$  179.2) at C-3 collapsed to a broad singlet and a broad doublet ( $^3J_{CH}=3.6$  Hz) coupled with a hydroxyl proton, respectively. Upon the same irradiation under the presence of  $D_2O$ , the signal at C-3 appears as a broad triplet coupled with the methylenic protons at C-9 and C-11. This spectral evidence means that the terminal carbon at C-5 in the butadienyl moiety, an ethyl group and an enolic group must be attached to the same quaternary carbon, which is bonded also to a sulfur atom. Further, the observation of a fermentation ion peak at  $m/z$  140 ( $C_8H_{12}S$ ) in the mass spectrum of **1** afforded the structural evidence for  $CH_2=CH-C(CH_3)=CH-C(CH_2-CH_3)-S-$ . Thus, we can propose the most suitable structure **1** for thiotetromycin. The validity of the structure as a 5-membered thiolactone was also supported by the spectroscopic characterization of two monomethyl ethers, **3** [ $\alpha$ ] $^{27}_D$  +63.5° ( $c$  1.0,  $CHCl_3$ ); UV  $\lambda_{max}^{EtOH}$  238 nm ( $\epsilon$  16,600); IR  $\nu_{max}^{CCl_4}$  1620  $cm^{-1}$  and  $^{13}C$  NMR  $\delta$  194.0 (thioester carbonyl), and **4**: [ $\alpha$ ] $^{27}_D$  +194.4° ( $c$  1.0,  $CHCl_3$ ); UV  $\lambda_{max}^{EtOH}$  235 ( $\epsilon$  16,100) and 313 nm (6,200); IR  $\nu_{max}^{CCl_4}$  1580  $cm^{-1}$  and  $^{13}C$

NMR  $\delta$  201.5 ( $\alpha,\beta$ -unsaturated ketone carbonyl), obtained by treatment of **1** with diazomethane. These spectral data demonstrated that **4** is an tautomeric isomer of **3**. The synthesis of **1** and its related compounds are now in progress.

#### Acknowledgment

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SATOSHI ŌMURA  
AKIRA NAKAGAWA  
RIMIKO IWATA  
AKIKO HATANO

School of Pharmaceutical Sciences,  
Kitasato University  
and The Kitasato Institute,  
Minato-ku, Tokyo 108,  
Japan

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